STEREOSPECIFIC SYNTHESIS OF (+)-BIOTIN

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In the course of our studies on the stereospecific syntheses of optically active Biotin Vitamers by use of the asymmetric carbons of carbohydrates, the syntheses of (+)-Oxybiotin and (+)-Dethiobiotin have been already achieved. We would now like to describe the stereospecific total synthesis of (+)-Biotin from D-mannose.

Treatment of 2,3:5,6-di-0-isopropylidene- α -D-mannofuranose(II)³ with benzoyl chloride in pyridine gave 1-0-benzoate(III)⁴[97%; mp 127-128°; $[\alpha]_D^{20}$ +33.4° (c 2.3, CHCl₃)]. Selective hydrolysis of the 5,6-0-isopropylidene group of III with 70% acetic acid at 20° for 48 hr gave 1-0-benzoy1-2,3-0-isopropylidene- α -D-mannofuranose(IV)[93%; glass; [α] $_{D}^{20}$ +52.5° (c 1.7, $C_{2}H_{5}OH$)]. Periodate oxidation of IV in a mixture of acetone and water afforded the aldehyde V, which was treated with excess [3-carbomethoxy-propen-(2)-ylidene-(1)]-triphenylphosphorane 5 in CH $_2$ Cl $_2$ to give methyl 1-0-benzoyl-2,3-0-isopropylidene-5,6,7,8-tetradeoxy- α -D- $\frac{1}{2}$ yxo-non-5,7-dieno- $\frac{1}{2}$,4furanuronate(VI)[5:6 trans 6:7 trans isomer; 85% from IV; mp 91-92°; $\lceil \alpha \rceil_D^{20} \rceil$ -33.6° (c 0.3, CHCl₃), 5:6 cis 6:7 trans isomer; 5%; mp 141-142°; $\lceil \alpha \rceil_D^{20} \rceil$ -120.5° (c 1.1, CHCl₃)]. Hydrogenation of IV with 10% Pd-C in methanol gave 1-0-benzoyl-2,3-0-isopropylidene-5,6,7,8-tetradeoxy- α -D- $\frac{1}{2}$ xo-1,4furanuronate(VII)[97%; mp 68-69°; $[\alpha]_D^{20}$ +30.0° (c 2.0, CHCl₃)]. Treatment of VII with NaOCH₃ in methanol, followed by the reduction of the resulting aldehyde VIII with $NaBH_4$ afforded methyl 2,3,4,5-tetradeoxy-7,8-0-isopropylidene-L- $\frac{1}{2}$ -nononate(IX)[85% from VII; syrup; $\left[\alpha\right]_{0}^{20}$ +12.3° (c 2.0, CHCl₃)]. Methanesulfonylation of IX gave a syrupy dimesylate X [96%; $[\alpha]_{D}^{20}$ +19.6° (c 3.0, CHCl₂)]. Treatment of X with sodium sulfide in hexamethylphosphoramide (HMPA) at 100° for 2 hr afforded a syrupy tetrahydrothiophene derivative XI [75%; $[\alpha]_D^{20}$ -91.2° (c 1.8, CHCl₃)]. Treatment of XI with 90% formic acid at 20° for 15 min gave a diol XII [92%; mp 55-56°; α] -108.6° (c 1.4, CHCl₃)]. Methanesulfonylation of XII gave a syrupy dimesylate XIII [95%; $\left[\alpha\right]_{n}^{20}$ -86.5° (c 2.1, CHCl₃)]. Treatment of XIII with sodium azide in HMPA at 80° for 7 hr afforded a syrupy diazido compound XIV [78%; $[\alpha]_D^{20}$ +42.1° (c 2.3, CHCl₃)]. Catalytic reduction of the azido groups of XIV with $Pt0_2$ in a mixture of methanol and acetic anhydride at 20° for 1 hr gave 2(S)-4'-carbomethoxy-buty1-1'-3(S)-azido-4(R)-acetamido-tetrahydrothiophene(XV)[73%; mp 115-116°; $\left[\alpha\right]_{0}^{20}$ -181.0° (c 1.0, CHCl₃); nmr (CDCl₃) δ 1.8 (6H, m, mehtylenes), 2.40 (3H, s, NCOCH₃), 2.34 (2H, t, J=6.5Hz, $CH_2CO_2CH_3$), 2.85 (2H, m, H_5), 3.61 (1H, m, H_2), 3.58 (3H, s, CO_2CH_3), 4.20 (1H, t, $J_{2,3}=J_{3,4}=4Hz$, H_3) 4.60 (1H, m, H_4) 6.75 (1H, d, J=8.0Hz, NH)]. Further reduction of XV under similar conditions for 3 hr afforded a diacetamido derivative XVI [80%; syrup; $[\alpha]_D^{20}$ -123.9° (c 1.7, C₂H₅OH)]. Compound XVI was obtained(60%) directly from XIV without the isolation of XV. Treatment of XVI with Ba(OH), in water at 140° for 14 hr, followed by the treatment with

REFERENCES AND FOOTNOTES

agreed with that of authentic I.

- 1. H. Ohrui, H. Kuzuhara and S. Emoto, Agr. Biol. Chem. (Tokyo), 35, 752 (1971).
- 2. H. Kuzuhara, H. Ohrui and S. Emoto, Agr. Biol. Chem. (Tokyo), 35, 8 (1971).
- 3. K. Freudenberg and A. Wolf, Ber., 60, 232 (1927).
- 4. Satisfactory elemental analyses, ir and nmr data were obtained for all compounds.
- 5. E. Buchta and F. Andree, Ber., 92, 3111 (1959).
- S. A. Harris, D. E. Wolf, R. Ralf, G. E. Arth, R. C. Anderson, N. R. Easton and K. Folkers, <u>J. Am. Chem. Soc.</u>, <u>67</u>, 2096 (1945).
- F. Kögl, H. Erxleben, J. H. Verneek, and W. A. J. Borg, Z. Phys: ol. Chem., 279, 121 (1943).